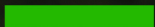




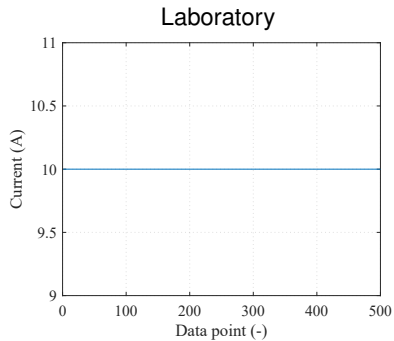
LAND OF THE CURIOUS



Linearization of alkaline electrolyzer current-voltage behavior under dynamic operation

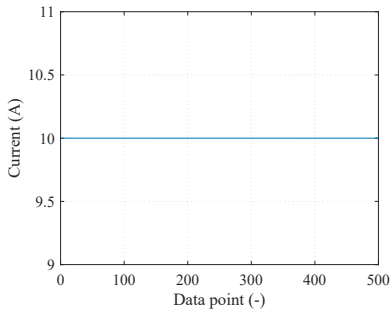


Why is this relevant?



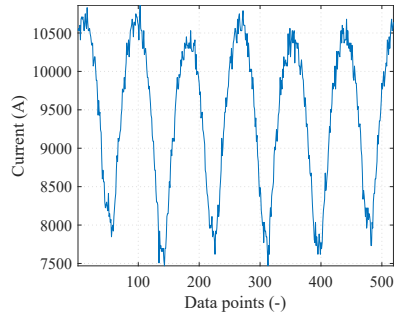
Why is this relevant?

Laboratory



→
Industrial
scale

Industry



Hydrogen production

- ▶ Hydrogen production is determined by the DC current

$$\dot{n}_{\text{H}_2} = \eta_F \frac{jA}{n_e F},$$

η_F : Faraday efficiency
 j : Current density
 A : Cell area

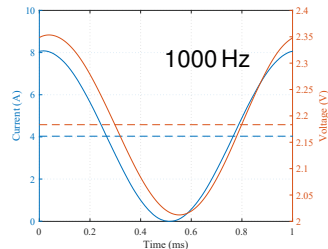
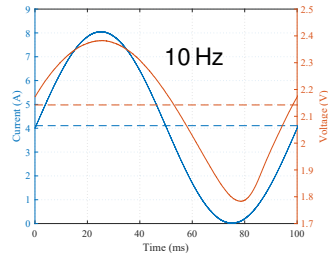
n_e : Transferred electrons
 F : Faraday constant

- ▶ Only direct current produces hydrogen
- ▶ AC current components transform to losses when supplied through the resistive electrolyzer load

Linearization of the electrolyzer behavior

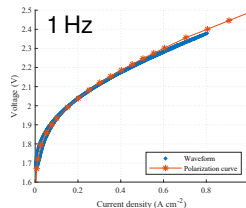
Ripple measurements

- ▶ Supply ripple to the electrolyzer cell
 - ▶ Sine waveform
 - ▶ 0-100% ripple amplitude
 - ▶ 0-1000 Hz ripple frequency
- ▶ Power analyzer used to measure current and voltage waveforms



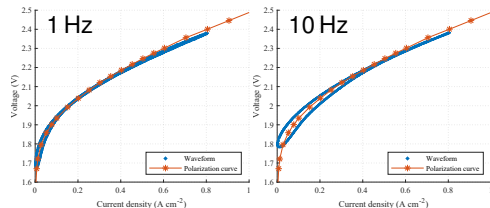
Electrolyzer cell voltage response at varying frequencies

- ▶ At low frequencies system mostly follows polarization curve as expected



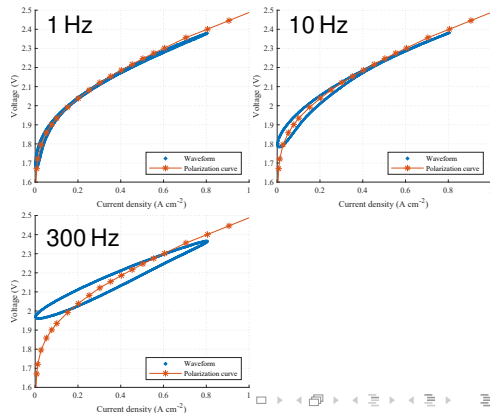
Electrolyzer cell voltage response at varying frequencies

- ▶ At low frequencies system mostly follows polarization curve as expected
- ▶ As frequency increases the waveform starts to linearize



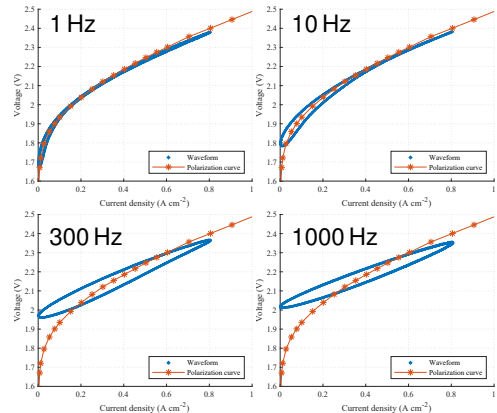
Electrolyzer cell voltage response at varying frequencies

- ▶ At low frequencies system mostly follows polarization curve as expected
- ▶ As frequency increases the waveform starts to linearize
- ▶ System mostly linear at 300 Hz

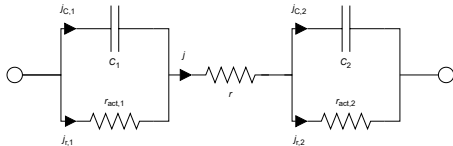


Electrolyzer cell voltage response at varying frequencies

- ▶ At low frequencies system mostly follows polarization curve as expected
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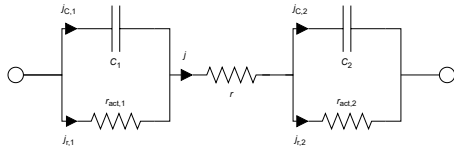


Why the linearization happens

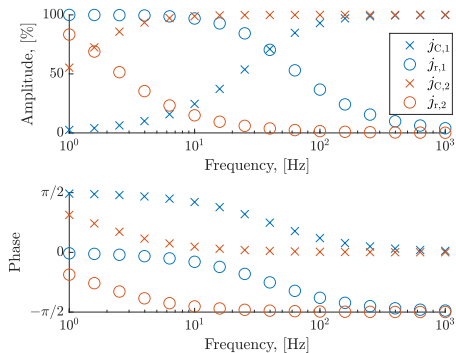


- ▶ Cell capacitances at the electrodes provide low impedance path for high frequency AC currents
- ▶ Amplitude of the AC current through the activation is reduced

Why the linearization happens

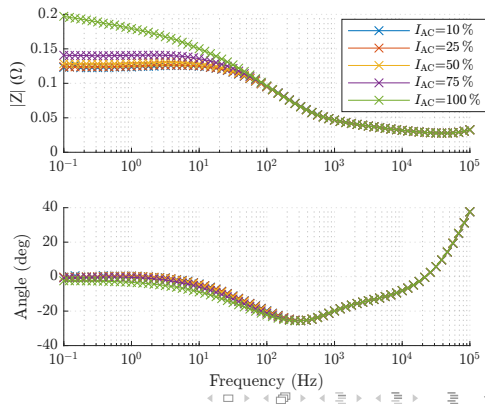


- ▶ Cell capacitances at the electrodes provide low impedance path for high frequency AC currents
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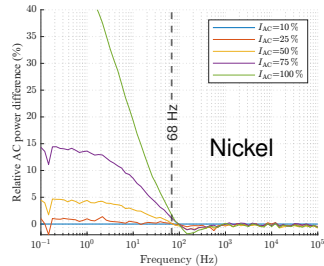
Defining the linearization frequency

- ▶ Nonlinear electrochemical impedance spectroscopy
 - ▶ Measurements done with a potentiostat
- ▶ Ripple amplitudes between 10-100% were used
- ▶ Convergence of the measurements indicates linearization



Finding the frequency limit for linearization

- ▶ Two electrode sets tested
 - ▶ Nickel foam
 - ▶ Advanced materials (DTU)
- ▶ System considered linear when relative AC power difference < 2% between all measurements



Equations for Frequency component I :

$$|Z_I| = \frac{U_I}{I_I}$$

$$P_{AC,I} = \frac{|Z_I| I_I^2}{2} \cos \theta_I$$

$$\theta_I = \phi_{U,I} - \phi_{I,I}$$

Z : Impedance
 I : Current

U : Voltage
 θ : Phase

Finding the frequency limit for linearization

- ▶ Two electrode sets tested
 - ▶ Nickel foam
 - ▶ Advanced materials (DTU)
- ▶ System considered linear when relative AC power difference $< 2\%$ between all measurements
- ▶ Advanced electrodes have higher surface area
 - higher capacitance
 - Lower linearization frequency

Equations for Frequency component I :

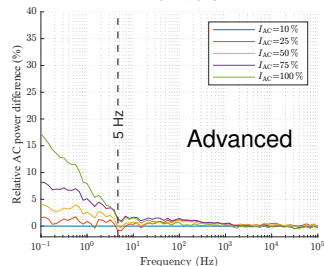
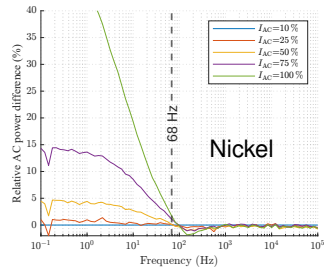
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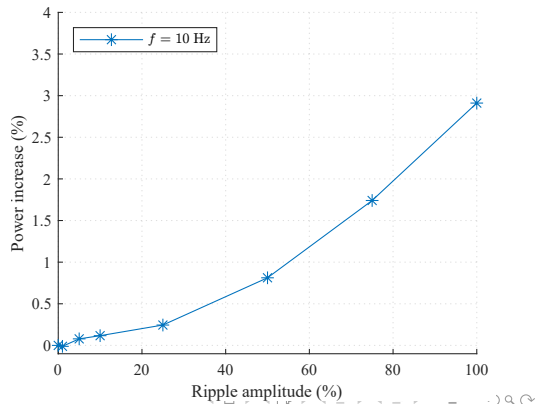
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Performance impacts of the dynamic operation

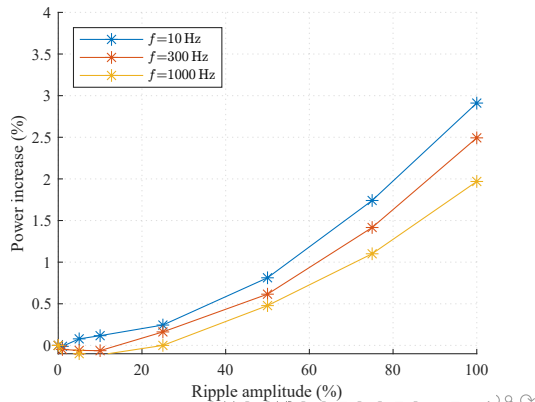
Additional power consumption

- ▶ Ripple amplitude increases the losses



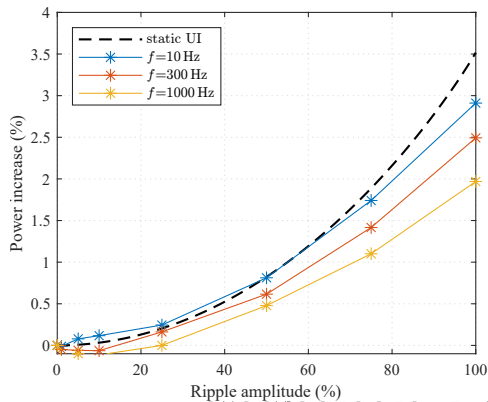
Additional power consumption

- ▶ Ripple amplitude increases the losses
- ▶ Ripple frequency counteracts the losses



Additional power consumption

- ▶ Ripple amplitude increases the losses
- ▶ Ripple frequency counteracts the losses
- ▶ As frequency is reduced losses approach the case where ripple follows the static UI curve
- ▶ Ripple following the static UI curve should present the maximum loss





Thank you